How can we sequester soil organic carbon?

Plant photosynthesis is the only process by which carbon is taken from the atmosphere and a fraction deposited in the soil through inputs of plant organic matter.

- Soil organic carbon input rates are determined by the root biomass of a plant, but also include stubble and leaf litter deposited from aboveground plant material.
- Practices that improve plant water use and growth (e.g. early sowing) are desirable because they also increase organic inputs into soil.

The capacity of a soil to store soil carbon over a long period of time is largely determined by the characteristics of that soil and climatic factors (this is referred to as 'attainable' soil carbon). Soils that have more clay content and occur in higher rainfall environments have been found to be able to store more carbon, while sandier soils in drier environments tend to be lower in soil carbon. Increasing the rate of organic inputs on coarse sandy soils may therefore not result in stable increases in soil organic carbon but may help to maintain the current soil carbon stock.

Soil management activities can be used to move soil carbon stocks towards their attainable levels. For example, in the Avon Arc

CEC indicates the potential CEC indicates the potential capacity of soil to store capacity of soil to store nutrients. The three main nutrients. The three main cations essential for plant cations essential for plant growth are potassium, growth are potassium, calcium and magnesium. calcium and magnesium. These influence soil These influence soil structure, colour and aggregate stability.

• soil structure:

soil organic carbon interacts with and influences the straining \mathbf{s} formation of soil structure, helping the formation of soil helping the formation of \mathcal{L}

region, maximum attainable carbon levels in cropping systems on sandy soils have been estimated to be approximately 40t-C/ha, while studies have measured an average actual carbon level of 19t-C/ha in this area.

Limiting gaseous emissions (respiration) of carbon from soils is a sequestration process. There are a number of ways to do this, the easiest to achieve are below:

- Limit soil disturbance to ensure the carbon protected from decomposition by soil microbes by clay or soil aggregates continues to be protected.
	- Increase plant cover to ensure there is an input of carbon to the soil from root and above-ground biomass. Soil left fallow is a net source of carbon to the atmosphere because there is no addition of carbon to counterbalance the loss of carbon from erosion or microbial respiration
- important for all key soil functions
- provides nutrients for example the majority of available soil nitrogen derived from soil organic matter comes from the humus fraction
- **• recalcitrant organic carbon** : biologically stable carbon, most common form is
- charcoal decomposes very
	- slowly and is therefore unavailable for use by micro-organisms
	- carbon that will not be readily-emitted to the atmosphere as CO₂

2 P_{max} **Page 3** P_{max} is the contract of P_{max} 3 **" more carbon = more plant available water holding capacity."**

Respiration rates are highest when conditions are warm and moist, meaning that summer rainfall can cause the rapid release of soil carbon, particularly if there are no active plants to replace the lost carbon. A recent Wheatbelt study by UWA suggested that soil organic carbon levels had, on average, dropped.

• water holding capacity in **• water holding capacity in soils:**

carbon acts like a sponge for soil water... more carbon = more plant available water. The more holding capacity. Although these increases may be small, they may be valuable in below average rainfall years.

Soil organic carbon in Wheatbelt cropping systems

- **Types of soil organic carbon and their role in agricultural soils:**
- **• crop residues:** above and below ground plant residues (leaves, stalks, roots) less than 2 mm long or wide
	- break down quickly
	- source of energy for soil biological processes
- **• particulate organic carbon** : plant residues that are smaller than 2 mm but

Soil organic carbon is Soil organic carbon is

important for: • cation exchange capacity important for: • Cation Exchange Capacity (CEC):

> larger than 0.053 mm • breaks down relatively

- quickly but more slowly than crop residues
- important for soil structure
- source of energy for biological processes source of nutrients

• humus decomposed materials :

less than 0.053 mm that are dominated by molecules stuck to soil minerals

What are the forms for soil organic carbon?

Why manage soil organic carbon?

By Jo Wheeler

Nutrient Use **Efficiency**

Poor timing of nutrient application and the loss of nutrients during intense weather events

Poor soil structure

Low soil biological activity.

An integrated approach to nutrient management supports landholders through soil analysis, development of nutrient management plans and better practices, improving crop performance through soil health while reducing nutrient export to waterways. This is a joint initiative between Perth NRM, Wheatbelt NRM and the Department of Parks and Wildlife.

utrient use efficiency is becoming increasing important as the cost of fertilisers continue to rise. Additional, Swan-Avon River Catchment modelling has identified the importance of addressing nutrient management on farms for healthy waterways **N**

The Nutrient Management Use Effieciency program is to developing soil monitoring, management and nutrient planning demonstration sites across the Avon River Basin which will then be used to develop a series of extension activities and materials to encourage neighbouring farmers to prioritise soil health and nutrient management.

This extension will build an understanding of the soil health factors limiting production amongst the farming community (Soil pH, compaction, salinity, toxicity, pests and diseases), and provide training in the use of decision support tools and technological advances that will enable farmers to manage nutrients better and to independently interpret soil testing results to guide soil management.

The Nutrient Use Efficency project process:

Farm nutrient-use inefficiencies are caused by:

Average annual nitrogen loads for the modelling catchments for the farm nutrient management scenarios*

Inefficient use of farm nutrients is causing lost profitability and poor river health." "

Average annual nitrogen reporting catchment loads for the farm nutrient management scenarios*

The table below explores the nutrient inputs and outputs for an average Wheatbelt wheat & sheep and mixed grazing farm based on current farming practices ('base case'), liming to address soil acidity and implementing on farm nutrient management. The table demonstrates that nutrient use efficiency (NUE) for nitrogen currently sits at about 41% and phosphorous at 48% in wheat & sheep farming. However, this can be improved to 48% (nitrogen) and 56% (phosphorous) by addressing soil acidity, or 56% (nitrogen) and 75% (phosphorous) through the adoption of relatively simple nutrient management practices. This equates to an average of almost 15 kg/ha/yr of nitrogen and 2.6 kg/ha/yr of phosphorous not being 'wasted' (surplus) on farm simply by liming and incorporating nutrient management practices such as soil testing for nutrient sufficiency to depth, plant tissue testing and better timing of fertiliser application. If all farmers in the Avon Basin were to adopt these improved nutrient management practices, the modelling suggests nitrogen loads would decrease by 153 t/yr and phosphorous loads by 2.4 t/yr at the catchment outlet, with the Mortlock, Dale and Middle/Upper Avon catchments providing the greatest decrease in nutrient out flows.

> **Average annual phorphorus reporting catchment loads for the farm nutrient management scenarios***

The major cause of nutrient pollution in the Avon Basin is the inefficient use of farm nutrients*. Farm nutrient-use inefficiencies are caused by:

By Dr Guy Boggs

*This article and the associated tables and graphs are based on information from: Hennig, K & Kelsey, P 2015, Avon Basin hydrological and nutrient modelling, Water Science Technical Series, report no. 74, Water Science Branch, Department of Water, Perth, Western Australia.